
All-aluminum liquid flow battery electrode reaction

Why do redox flow batteries have porous electrodes?

Porous electrodes are critical in determining the power density and energy efficiency of redox flow batteries. These electrodes serve as platforms for mesoscopic flow, microscopic ion diffusion, and interfacial electrochemical reactions.

How do redox-active batteries work?

Fundamentally, they have adopted electrode designs from conventional rigid batteries that rely on the mechanical coupling (solid-to-solid contact) of the redox-active species and the conductive filler (Fig. 1A). A coupled electrode requires a binder to provide mechanical integrity in the solid matrix by holding the respective components together.

What are aqueous aluminum-ion batteries?

In recent years, aqueous aluminum-ion batteries (AAIBs, hereafter) have become an essential direction for materials science and engineering research. (1-4) Compared to traditional lithium-ion batteries, AAIBs have many advantages regarding being inexpensive, having high safety, and being abundant.

Why do redox-active electrofluid batteries have a lower volumetric capacity?

Furthermore, thicker electrodes tend to have higher electrical resistance and tortuosity that hinder electrical and ion transport, limiting access to the active species in the solid electrode, resulting in a lower effective volumetric capacity (11,12). Fig. 1. Redox-active electrofluid stretchable battery concept.

Aqueous aluminum-ion batteries (AAIBs) are attractive electrochemical cells for energy storage because of Earth's crust abundance, inexpensiveness, high theoretical ...

Vanadium redox flow batteries (VRFBs) are considered as promising electrochemical energy storage systems due to their efficiency, flexibility and scalability to meet our needs in ...

Electrode materials, in particular, play a crucial role in battery performance, as their redox reactions involve the simultaneous accommodation and release of electrons and charge ...

Electrodes, which offer sites for mass transfer and redox reactions, play a crucial role in determining the energy efficiencies and power densities of redox flow batteries. This ...

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Here, the authors report an organic self-charging flow battery that charges within 8 minutes to 94% capacity, matches various multivalent metal negative electrodes, and ...

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AlCl₃/[EMIm]Cl ionic liquid Figure 1. Comparison of theoretical specific capacities [1] of different anode materials (a); photo of as-prepared rechargeable battery (b) using liquid ...

seful" surface area is dictated by rate of reaction and transport within the electrode. Finally, I investigate the viability of nickel metal electrodeposition on carbon electrodes to ...

After the electrochemical reaction through the electrode, the electrolyte returns to the liquid storage tank and circulates in turn to realize the charge and discharge of the battery ...

Achieving cathodes with large areal capacities is crucial for advancing aqueous aluminum-based batteries. Here, authors report a hydrate-melt electrolyte based on AlCl₃ and ...

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